



Research Article

DISCRIMINATION OF *DAEDALEOPSIS NITIDA* MUSHROOMS THAT GROWING IN DIFFERENT ENVIRONMENTS USING FOURIER TRANSFORM INFRARED SPECTROSCOPY

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ABSTRACT

In this study, the chemical structure analysis of two *Daedaleopsis nitida* mushrooms that grown in two different environments have been investigated. Fourier Transform Infrared Spectroscopy (FT-IR) was used to determine functional group vibrations in the chemical contents of two fungi species. As a result of the investigation, it was observed that there were small chemical contents in the structure of two mushrooms belonging to the same species.

Keywords: FT-IR, *Daedaleopsis nitida*, mushrooms.

1. INTRODUCTION

Mushrooms are very interested by people all over the world because they are low energy level and high in vegetable proteins, important vitamins, iron, chitin, fibre and some minerals [1, 2, 3, 4, 5]. The value of mushroom in terms of tonic and medicinal is used in the folk medicinal for thousands of years [6, 7, 8]. Mushroom is used in the treatment of many diseases as a very good source of nutraceuticals, antioxidants, anticancerous, antitumor, immune-modulating, antiinflammatory, cardiovascular, enzyme inhibitory activities, antimicrobial and antidiabetic agents [9, 10, 11]. Mushrooms is using to decrease pollution levels in a given area [12].

Structural analysis can be applied when the areas of use are determined. It is used in spectral analysis methods such as FT-IR spectra [2, 3, 10, 12, 13, 14], second derivative spectra [14], NMR spectroscopy [3, 13], GPC and HPSEC-MALLS analyzes [15], while methods such as Microscopic image of tissue section [16], Principal component analysis [2, 17], hierarchical cluster analysis [18] and curve-fitting analysis [19].

But the FT-IR spectrum is the most widely used method to determine the functional groups [2, 3, 10, 12, 13, 14] as we did so. In the literature [12, 20] we knew from these studies that mushroom absorbs the chemicals and pollution in the environment because of its absorbent

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property. There is no such study in the literature that analyzes two samples of the same species using FT-IR spectroscopy. With this study, the literature will gain a new perspective.

2. MATERIALS AND METHODS

2.1. Samples Collection

Two fungi (mushroom A and B) belonging to *Daedaleopsis nitida* (Durieu & Mont.) Zmitr. & Malysheva (Syn: *Apoxona nitida* (Durieu & Mont.) Donk; *Hexagonia nitida* Durieu & Mont.) were collected from different localities. In the field, the specimens were photographed on the substrates. The morphological and ecological characteristics were recorded. After the field studies, the specimens were taken to the laboratory. Micromorphological characters were observed by light microscope using KOH, Melzer's reagent, Congo red and distilled water. Identification of the samples was conducted according to Marchand [21]. The dried samples were conserved at the fungarium in the Süleyman Demirel University. Examined examples; **A**: Turkey, Balıkesir, Kazdağları national park, on oak wood, 26.10.2013, leg. & det. Ö.F. Çolak (ÖFÇ 206), **B**: Turkey, Burdur, Bucak district, Kargı village, Sweetgum forest protected area, on sweetgum wood, 11.05.2017, leg. & det. Ö.F. Çolak (ÖFÇ 1292).

2.2. Infrared Analysis (FT-IR)

The main equipment used was Spectrum BX FT-IR spectrometer system (Perkin Elmer). The Infrared spectra were recorded by accumulating 16 scans each in 4000-400 cm^{-1} range with a resolution of 4 cm^{-1} with KBr pellets at room temperature.

2.3. Procedure

About 1 mg *Daedaleopsis nitida* samples powder was mixed evenly with 100 mg KBr. KBr pellets were prepared exerting pressure of not more than 10 psi for approximately 2 min in a pellet press. The FT-IR spectrum of pellet was usually acceptable when a transmission of 60-70% was accomplished. To eliminate any prevention that might be caused by variation in pellet thickness, the test had to be repeated with either same sample powder or with more KBr added.

3. RESULTS AND DISCUSSION

3.1. Data analysis for FT-IR spectra

The FT-IR spectrums of *Daedaleopsis nitida* harvested from different areas are shown of samples **A** and **B** in Figure 1. The vibrational spectroscopic results for mushrooms are given in Table 1.

In FT-IR spectra of mushrooms a broad band between 3600 and 3200 cm^{-1} is O-H stretching band [18, 22] and the peaks located at 2925-2921 cm^{-1} in **A**, **B**, respectively, corresponds to CH_2 asymmetric stretching vibration in sugar residue [23]. Absorption band about 2853 cm^{-1} (shoulder) correspond to symmetric stretching vibration of methylene (CH_2) group [18].

When the peak at 2288 cm^{-1} is emerged in **B**, it wasn't observed in **A**. It should be due to two different nitrogen base like N-terminal amino acids in peptides and also due to purines and pyrimidine bases in nucleotides [23].

The peaks which indicate the presence of protein in **A** and **B** were observed with absorption at approximately 1651, 1623 and 1654 cm^{-1} for amide I, 1513 and 1511 cm^{-1} for amide II. They represent C=O groups in the FT-IR spectra of mushrooms for amide I [2, 22]. The amide II absorption was attributed to N-H bending vibration coupled to C-N stretching vibration mode of the polypeptide protein backbone [18]. As a result of the **B** bifurcation, the degradation of the structure originating from the environment can be considered.

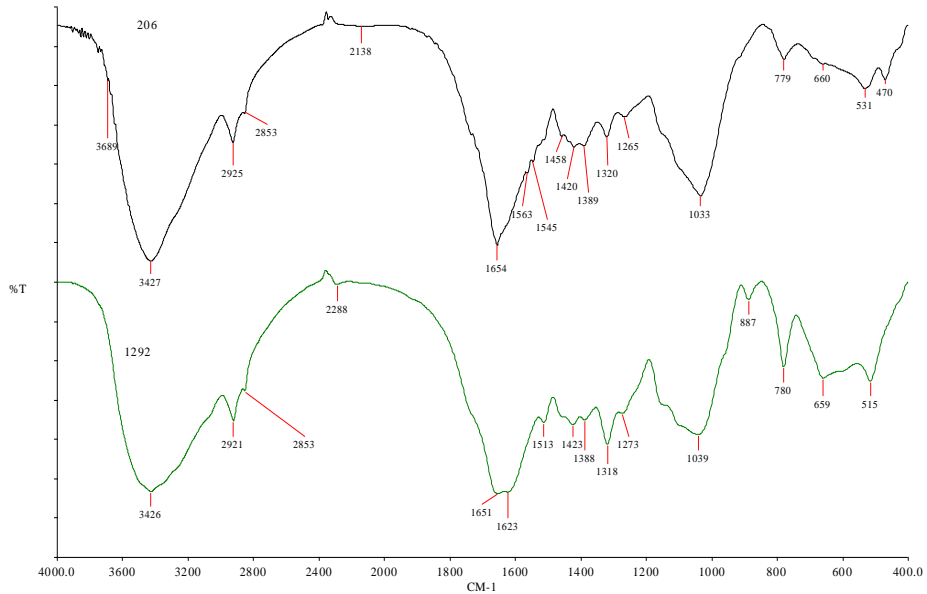


Figure 1. The infrared spectra of *Daedaleopsis nitida* harvested from different areas for samples A (206) and B (1292).

Table 1. Comparison of the main peaks assignment of two samples of *Daedaleopsis nitida* from different locations.

Band (cm ⁻¹)	Vibration mode	Mushrooms	
		A	B
3465-3200	-OH stretching	3427(br)	3426
2990-2855	Asymmetric CH ₂ stretching	2925	2921
2990-2855	Symmetric stretching methylene (CH ₂) group	2853(sh)	2853(sh)
2310-2000	C=NH ₂ , C=N ⁺	-	2288
1800-1630	Amid I group [2]	1654	1651-1623
1510-1460	Amid II group [2]	1511	1513
1480-1405	Asymmetric CH ₃ bending	1458	1454
1433	CO-NH, C=C	1420	1423
1379	Symmetric CH ₃	1389	1388
1310	C-O-H bending and CH ₂ deformation	1320	1318
1240	C-N stretching and N-H deformation	1265	1273
1041	Symmetric =C-O-C, in plane bending-CH ₂ OH	1033	1039
887	C=CH ₂	-	887
776	C-H out of plane bending	779	780
691-616	O-N=O bending	660w	659w
531	S-S stretching	531-470w	-
690-515	C-Br stretching	-	515

sh:shoulder, br:broad, w:weak

The peaks located at 1454,1458 cm^{-1} were due to asymmetric CH_3 bending mode of protein methyl group, 1423,1420 cm^{-1} are assigned to phosphates in nucleotides and bending modes of CO-NH group in the **A**, **B** in CH_2 group [23].

The bands centered at 1388-1389 cm^{-1} , respectively **B** and **A**, were assigned to symmetric CH_3 vibration mode [14]. The absorption band at around 1320-1318 cm^{-1} , respectively **A** and **B**, mainly originated from C-O-H bending and CH_2 deformation vibration [18].

The peaks 1273 (in **B**) -1265 (in **A**) cm^{-1} were mostly due to the amide III band components of protein, corresponding to C-N stretching vibrations and N-H deformations vibration [18, 23].

The bands centered at 1039 (in **B**) -1033 (in **A**) cm^{-1} corresponds to symmetric stretching of =C-O-C, and symmetric in plane bending - CH_2OH in polysaccharides [14, 18].

The medium band at around 780-779 cm^{-1} in **B**, **A** in respectively belongs to the characteristic absorption of β -glucan, corresponding to C-H out of plane bending [18, 24]. In the above order, the weak bands observed at 659-660 cm^{-1} due to O-N=O bending shows the presence of nitrite [24].

The peak 887 cm^{-1} (only in **B**) is probably attributed to the $\text{C}=\text{CH}_2$ presented that it is mostly assigned to the existence of polysaccharides, such as β -D-Glucan, the pyranose form of glucose [25].

The weak absorption bands observed at 515 cm^{-1} only in **B** due to C-Br stretching show the presence of Bromine [26].

The weak absorption bands observed at 531 and 470 cm^{-1} only in **A** due to S-S stretching show the presence of sulphide [24, 27].

After detailed examination of the FT-IR spectroscopy, little chemical changes were observed in the structure of two fungi grown in different area of the same species. These small differences are an indication that fungi absorb small molecules in the environment. In the meantime, it is useful to add that there must be molecules that can fit into the holes within our own chemical structure. This shows us that fungi can also be used as chemical sieves at the same time.

4. CONCLUSION

We have investigated that chemical analysis of two *Daedaleopsis nitida* mushrooms belonging to the same species in two different environments. FT-IR spectroscopy was used for this research. As a result of the studies carried out, it was determined that there are minor changes in the chemical structure of the mushrooms grown in different environments of the same species. These small changes showed; even if they have the same species, the fungicidal structure of the fungi may show changes depending on the environment in which they grow. We generally encounter these changes in the area of 1500-400 cm^{-1} which we call fingerprint region in FT-IR spectroscopy. These results mean that the mushrooms can trap small chemicals with the effect of outer space. This event can also create the impression that the fungi are biological sifter as well as a lot of other benefits that can be used for environmental cleansing.

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